

Interconversion of *cis*- and *trans*-Fused Oxabicyclo[5.2.0]nonan-2-ones

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Dedicated to Professor Kurt Schaffner on the occasion of his 80th birthday

On irradiation (350 nm) in the presence of alkenes (2,3-dimethylbut-2-ene, 1,1-dimethoxyethene, and 2,4,4-trimethylpent-1-ene), benzoxepinone **1** and dioxepinone **2** are converted into mixtures of *cis*- and *trans*-fused oxabicyclo[5.2.0]nonan-2-ones. Their relative thermodynamic stabilities (as reflected by the observed diastereoisomer ratios after equilibration with basic alumina) depend on the substitution pattern of the alkene moiety.

Introduction. – Stepwise [2 + 2] photocycloaddition of five- or six-membered cyclic enones to alkenes to afford bicyclo[3.2.0]heptan-2-ones or bicyclo[4.2.0]octan-2-ones represents one of the most commonly used preparative light-induced reactions [1][2]. Whereas, for the former bicycles, only *cis*-fused examples are known, the latter exist in both forms, although the *trans*-fused diastereoisomers are converted quantitatively to the *cis*-bicycles under basic conditions. No similar approach to the homologous bicyclo[5.2.0]nonan-2-ones has been reported in the literature, most probably due to the fact that, on irradiation, cyclohept-2-enones undergo efficient (*Z/E*)-isomerization, followed by (thermal) dimerization of the so formed (*E*)-diastereoisomers [3–5]. Interestingly, calculations on such 7/4 bicycles suggest that both diastereoisomers should be of approximately equal energy [6][7]. Very recently, we have reported that seven-membered cyclic oxanones undergo [2 + 2] photocycloaddition to 2,3-dimethylbuta-1,3-diene efficiently [8]. Here, we report on the reactions of (excited) cyclic oxanones **1–3** in the presence of alkenes **4–6**, respectively, and on the subsequent base-induced equilibration of the resulting photocycloadducts (*Fig.*).

Results. – Irradiation of benzoxepinone **1** in the presence of a tenfold molar excess of 2,3-dimethylbut-2-ene (**4**) in benzene as solvent afforded a 1 : 5 : 5 mixture of adduct **7**, and of cycloadducts *cis*-**8** and *trans*-**8** (40% overall) in addition to tricyclic photodimers of **1**. Isolation of the 1 : 1 mixture *cis*-**8**/*trans*-**8** was achieved by chromatography. Irradiation of **1** in the presence of 1,1-dimethoxyethene (**5**) afforded a 2 : 3 mixture of cycloadducts, *cis*-**9**/*trans*-**9** (90% overall), with only traces of dimers of **1** being detected. Irradiation of **1** in the presence of 2,4,4-trimethylpent-1-ene (**6**) gave a 25 : 9 : 15 : 30 : 21 mixture **10**/*cis*-**11**/*cis*-**12**/*trans*-**11**/*trans*-**12** (70% overall) as monitored by ¹H-NMR spectroscopy, again in addition to photodimers of **1** (*Scheme 1*).

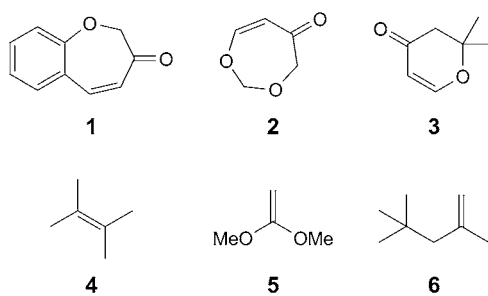
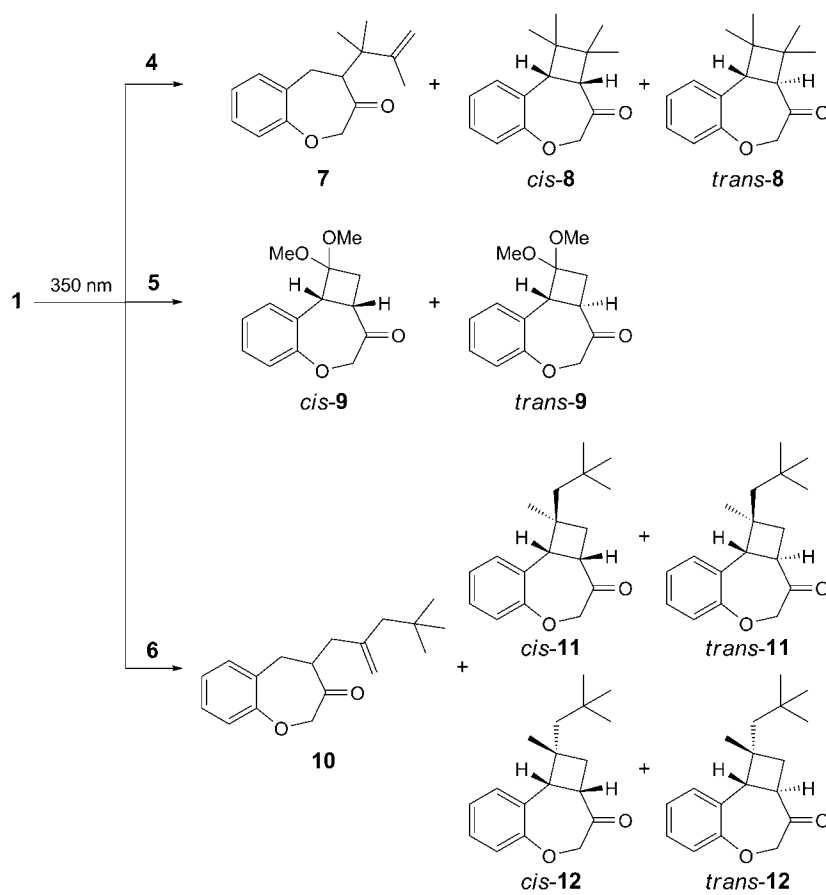


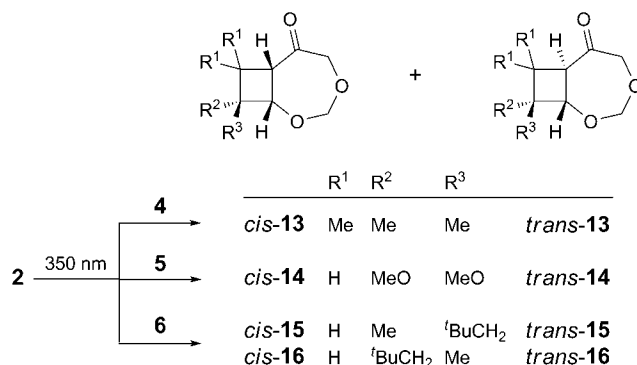
Figure. Starting materials for the described reactions

Scheme 1

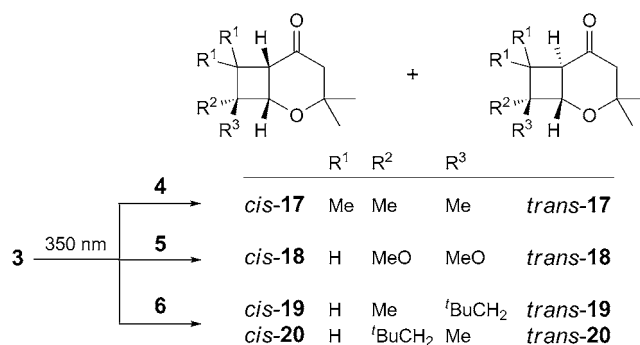


Similarly, irradiation of dioxepinone **2** in the presence of **4** afforded a 2 : 3 mixture of cycloadducts *cis*-**13**/*trans*-**13**, while, in the presence of **5**, a 1 : 4 mixture *cis*-**14**/*trans*-**14**

was formed. From **2** and **6**, a 20:5:70:5 mixture *cis*-**15**/*cis*-**16**/*trans*-**15**/*trans*-**16** was obtained. In contrast to the reactions with **1**, no dimers were formed in these irradiations of **2**, the formation of mixed cycloadducts, therefore, being quantitative (*Scheme 2*).

Scheme 2

Finally, irradiation of pyranone **3** in the presence of **4** gave a 3:2 cycloadduct mixture *cis*-**17**/*trans*-**17**, while, in the presence of **5**, a 1:1 mixture *cis*-**18**/*trans*-**18** was obtained [9]. Irradiation in the presence of **6** yielded a 13:27:45:15 cycloadduct mixture *cis*-**19**/*cis*-**20**/*trans*-**19**/*trans*-**20**, respectively (*Scheme 3*).

Scheme 3

Equilibration of the so formed mixtures of diastereoisomeric cycloadducts with basic alumina in CH₂Cl₂, as monitored by ¹H-NMR spectroscopy, proceeded in a diversified manner. As expected, all *trans*-fused oxabicyclooctanones, *trans*-**17**, *trans*-**18**, *trans*-**19**, and *trans*-**20**, underwent isomerization quantitatively to the corresponding *cis*-fused diastereoisomers. Regarding the oxabicyclononanones, both dimethoxyethene *trans*-cycloadducts, *trans*-**9** and *trans*-**14**, again isomerized quantitatively to the corresponding *cis*-fused diastereoisomers, respectively. In contrast, the tetramethylethene cycloadducts exhibited a different behavior, as *cis*-**8** isomerized nearly quantitatively to *trans*-**8**, whereas the diastereoisomer ratio *cis*-**13**/*trans*-**13** remained

almost unaffected changing from 2 : 3 to 1 : 1. Finally, for the cycloadducts of alkene **6**, a partial but not quantitative isomerization of all *trans*-fused diastereoisomers to the corresponding *cis*-fused cycloadducts was observed (Table).

Table. Percentages of *trans*-Fused Diastereoisomers after Equilibration with Basic Alumina in CH_2Cl_2 (reaction of enones **1–3** with alkenes **4–6**, resp.)

	4	5	6 (<i>anti</i>)	6 (<i>syn</i>)
1	8 (85%)	9 (0%)	11 (40%)	12 (40%)
2	13 (50%)	14 (0%)	15 (10%)	16 (40%)
3	17 (0%)	18 (0%)	19 (0%)	20 (0%)

Discussion. – The replacement of C- by O-atoms in a seven-membered cyclic enone increases the ring rigidity due to the shorter C–O vs. C–C bonds and thus slows down the relaxation of the triplet enone by twisting around the C=C bond, allowing its partial trapping by alkenes. Photocycloadducts *trans*-**8**, *cis*-**9**, and *cis*-**14** represent the first examples of isolated and purified, both *cis*- and *trans*-fused, bicyclo[5.2.0]nonan-6-one derivatives¹⁾. Differentiation between these diastereoisomers stems from both ¹H- and ¹³C-NMR spectra. Regarding the cycloadducts of **2**, the acetal CH₂ group represents an excellent indicator, both *via* the geminal H,H-coupling constant (²*J* of *ca.* 5 for *cis*-fused, and *ca.* 7.5 Hz for *trans*-fused diastereoisomers), and the difference in chemical shifts of the H-atoms (<0.1 for *cis*-fused, and ≥0.8 ppm for *trans*-fused diastereoisomers), reflecting the corresponding preferential conformation of the dioxepane ring [10]. In addition, the chemical shifts of the acetal C-atoms also differ distinctly (*δ ca.* 95 for *cis*-fused, and *ca.* 100 ppm for *trans*-fused diastereoisomers). In contrast, the differentiation between the corresponding cycloadducts of **1** is more subtle, the best indicators being the vicinal H,H-coupling constant of the bridgehead H-atoms (³*J* of *ca.* 11 for *cis*-fused, and *ca.* 13 Hz for *trans*-fused diastereoisomers), as well as the chemical shifts of the CH₂O C-atoms (*δ* ≥ 80 for *cis*-fused, and ≤ 79 ppm for *trans*-fused diastereoisomers). Finally the differentiation of diastereoisomeric oxabicyclooctanones **17–20** is straightforward following the indicators established before [11]. Adducts **7** and **10** exhibit spectroscopic data very similar to those for the (known) 4-methyl derivative [12].

Interestingly, alkene **6**, which we employed as higher-boiling substitute for (gaseous) isobutene, has so far only been reported twice as reactant in enone + alkene cycloadditions [13][14], although the differentiation between the resulting *syn*- and *anti*-diastereoisomers was not possible at that time. Here, in addition to NOE measurements, the chemical shifts of the neopentyl CH₂ C-atoms represents an excellent indicator (*δ* ≥ 55 for *anti*-cycloadducts and ≤ 50 ppm for *syn*-cycloadducts, resp.).

Whereas calculations on the stability relationship of diastereoisomeric bicyclic ketones such as bicyclo[3.3.0]octanones and bicyclo[4.3.0]nonanones have been

¹⁾ No references were found for the parent bicycle although a CAS registry is present (105104-52-7).

performed recently [15], only little corresponding information exists for bicyclo[5.2.0]- or oxabicyclo[5.2.0]nonanones, except, as already mentioned in the *Introduction*, for the assumption that here *cis*- and *trans*-fused diastereoisomers should be of approximately equal energy [6][7]. The experimental isomerization conditions used here were optimized (complete equilibration without decomposition of a diastereoisomer) with oxabicyclooctanone mixtures *cis*-**17**/*trans*-**17** – *cis*-**20**/*trans*-**20**.

Our results (*cf.* Table) do confirm this hypothesis, as indeed *both diastereoisomers are present after equilibration* for all 9,9-dialkyl- or 8,8,9,9-tetraalkyl-substituted oxabicyclo[5.2.0]nonan-2-ones, and, surprisingly, for compounds **8**, *i.e.*, **1** + **4** cycloadducts, the *trans*-fused diastereoisomer is the one of higher stability. The observation that the corresponding dimethoxy-substituted *trans*-fused diastereoisomers *trans*-**9** and *trans*-**14** do again quantitatively isomerize to the corresponding *cis*-diastereoisomers could be due to the fact that here the anomeric effect [16] represents a more important factor than conformational strain.

Experimental Part

1. *General*. Photolyses were conducted in a Rayonet RPR-100 photoreactor equipped with 350-nm lamps, and with solvents of spectrophotometric grade. Column chromatography (CC): silica gel 60 (SiO₂; Merck; 230–400 mesh). ¹H- and ¹³C-NMR spectra (including 2D plots): in CDCl₃; Bruker WM-500; at 500.13 and 125.8 MHz, resp.; δ in ppm rel. to Me₄Si as internal standard, *J* in Hz. GC/EI-MS: Varian MAT-311A at 70 eV.

2. *Starting Materials*. 1-Benzoxepin-3(2H)-one (**1**) was synthesized according to [17] and 2,3-dihydro-2,2-dimethyl-4H-pyran-4-one (**3**) according to [18]. 1,3-Dioxepin-5(4H)-one (**2**) was obtained by oxidation of the corresponding allylic alcohol [19] using pyridinium dichromate (PDC) as oxidant in DMF as solvent [20] instead of MnO₂ [19]. The alkenes **4**–**6** were commercially available. Basic alumina (pH 8.0; Woelm).

2.1. *Synthesis of 2*. A mixture of 10.6 g of PDC and 2.32 g (20 mmol) of 4,5-dihydro-5-hydroxy-1,3-dioxepine in DMF (40 ml) was stirred for 30 min at 0°, then for 6 h at r.t., poured into H₂O (500 ml), and then extracted (5 × 50 ml) with Et₂O/pentane 2 : 1. The combined org. layers were dried (MgSO₄), and the solvent was distilled off using a Vigreux column. The residue (600 mg, 22%), consisted of **2** (80%) and DMF (20%), and was used as such for the irradiations described below. Pure (> 95%) **2** could only be obtained by prep. GC.

3. *Photocycloadditions*. Ar Degassed solns. of enone **1**, **2**, or **3** (1 mmol) and alkene **4**, **5**, or **6** (10 mmol) in benzene (10 ml) were irradiated for 4–14 h until total conversion (monitoring by GC). After evaporation of the solvent and excess alkene, the residue was subjected to CC (SiO₂).

3.1. *Irradiation of 1 in the Presence of 4*. A soln. of **1** (160 mg) and **4** (840 mg) was irradiated for 14 h. CC (pentane/Et₂O 4 : 1) afforded first 15 mg (6%) of 4,5-dihydro-4-(1,1,2-trimethylprop-2-en-1-yl)-1-benzoxepin-3(2H)-one (**7**; *R*_f 0.51). Colorless liquid. ¹H-NMR: 7.35–7.05 (*m*, 4 H); 4.87 (*s*, 2 H); 4.44 (*s*, 2 H); 3.61 (*dd*, *J* = 3.6, 12.4, 1 H); 2.95 (*dd*, *J* = 12.4, 16.4, 1 H); 2.83 (*dd*, *J* = 3.6, 16.4, 1 H); 1.72, 1.28, 1.25 (3*s*, 9 H). ¹³C-NMR: 213.1 (*s*); 159.9 (*s*); 150.1 (*s*); 130.3 (*s*); 128.7 (*d*); 128.6 (*d*); 127.9 (*d*); 126.3 (*d*); 111.7 (*t*); 79.1 (*t*); 52.1 (*d*); 40.7 (*s*); 31.7 (*t*); 25.5 (*q*); 22.2 (*q*); 19.9 (*q*).

The second fraction (63 mg, 27%, *R*_f 0.41) consisted of a 1 : 1 mixture *cis*/*trans*-1,2,2a,9b-tetrahydro-1,1,2,2-tetramethylcyclobuta[d][1]benzoxepin-3(4H)-one (*cis*-**8**/*trans*-**8**). The isolation of *trans*-**8** is described in Sect. 4.1; the following data for *cis*-**8** stem from the oily product mixture: ¹H-NMR: 7.35–7.05 (*m*, 4 H); 4.40, 4.27 (*AB*, *J* = 17.2, 2 H); 3.87 (*d*, *J* = 10.9, 1 H); 3.38 (*d*, *J* = 10.9, 1 H); 1.30, 1.20, 1.18, 0.88 (4*s*, 12 H). ¹³C-NMR: 214.1 (*s*); 160.0 (*s*); 131.2 (*s*); 128.1 (*d*); 127.9 (*d*); 127.2 (*d*); 126.6 (*d*); 79.9 (*t*); 56.1 (*d*); 46.1 (*d*); 41.2 (*s*); 41.1 (*s*); 27.0 (*q*); 24.9 (*q*); 22.1 (*q*); 21.9 (*q*).

3.2. *Irradiation of 1 in the Presence of 5*. A soln. of **1** (160 mg) and **5** (880 mg) was irradiated for 14 h. CC (pentane/Et₂O 1 : 1) afforded 205 mg (80%) of a 2 : 3 mixture *cis*/*trans* 1,2,2a,9b-tetrahydro-1,1-

dimethoxycyclobuta[d][1]benzoxepin-3(4H)-one (*cis-9/trans-9*; R_f 0.48). The isolation of *cis-9* is described in Sect. 4.2; the following data for *trans-9* stem from the oily product mixture: $^1\text{H-NMR}$: 7.35–7.05 (*m*, 4 H); 4.62, 4.48 (*AB*, $J = 16.6$, 2 H); 3.65 (*m*, 2 H); 3.40, 3.28 (2*s*, 6 H); 2.40 (*dd*, $J = 7.8$, 12.0, 1 H); 2.35 (*dd*, $J = 9.6$, 12.0, 1 H). $^{13}\text{C-NMR}$: 209.1 (*s*); 159.9 (*s*); 130.3 (*s*); 128.7 (*d*); 128.6 (*d*); 127.9 (*d*); 126.3 (*d*); 101.3 (*s*); 78.0 (*t*); 50.5 (*d*); 50.4 (*q*); 50.2 (*q*); 42.5 (*d*); 30.2 (*t*).

3.3. Irradiation of **1** in the Presence of **6**. A soln. of **1** (160 mg) and **6** (1120 mg) was irradiated for 14 h. CC (pentane/Et₂O 7:1) afforded first 50 mg (19%) of *4,5-dihydro-4-(4,4-dimethyl-2-methylidenepentyl)-1-benzoxepin-3(2H)-one* (**10**; R_f 0.54). Colorless liquid. $^1\text{H-NMR}$: 7.35–7.05 (*m*, 4 H); 4.90 (*s*, 1 H); 4.80 (*s*, 1 H); 4.58, 4.42 (*AB*, $J = 17.2$, 2 H); 3.67 (*m*, 1 H); 3.12 (*dd*, $J = 4.1$, 17.0, 1 H); 2.75 (*dd*, $J = 12.0$, 17.0, 1 H); 2.67 (*dd*, $J = 4.9$, 15.2, 1 H); 2.06 (*dd*, $J = 9.1$, 15.2, 1 H); 1.94 (*s*, 2 H); 0.93 (*s*, 9 H). $^{13}\text{C-NMR}$: 213.9 (*s*); 159.9 (*s*); 146.1 (*s*); 130.3 (*s*); 128.7 (*d*); 128.6 (*d*); 127.9 (*d*); 126.3 (*d*); 117.1 (*t*); 79.1 (*t*); 50.1 (*t*); 47.3 (*d*); 37.1 (*t*); 32.9 (*t*); 30.9 (*s*); 29.6 (*q*).

The second fraction (133 mg, 47%; R_f 0.41) consisted of a 1:2:4:2 diastereoisomer mixture of *1-(2,2-dimethylpropyl)-1,2,2a,9b-tetrahydro-1-methyl-cyclobuta[d][1]benzoxepin-3(4H)-one* (*cis-11/cis-12/trans-11/trans-12*). The following data stem from the product mixture (aromatic H- and C-atoms not indicated).

cis-11: $^1\text{H-NMR}$: 4.60, 4.35 (*AB*, $J = 17.4$, 2 H); 3.85 (*d*, $J = 11.4$, 1 H); 3.80 (*m*, 1 H); 2.58 (*dd*, $J = 3.3$, 11.9, 1 H); 2.30 (*dd*, $J = 9.8$, 11.9, 1 H); 1.85, 1.75 (*AB*, $J = 14.1$, 2 H); 1.49 (*s*, 3 H); 1.02 (*s*, 9 H). $^{13}\text{C-NMR}$: 82.0 (*t*); 57.1 (*t*); 51.0 (*d*); 47.1 (*d*); 42.1 (*s*); 34.2 (*t*); 30.5 (*s*); 29.4 (*q*); 26.4 (*q*).

cis-12: $^1\text{H-NMR}$: 4.67, 4.35 (*AB*, $J = 17.4$, 2 H); 3.83 (*d*, $J = 11.4$, 1 H); 3.80 (*m*, 1 H); 2.33 (*dd*, $J = 3.3$, 11.9, 1 H); 1.97 (*dd*, $J = 9.8$, 11.9, 1 H); 1.30, 0.92 (*AB*, $J = 14.1$, 2 H); 1.27 (*s*, 3 H); 0.84 (*s*, 9 H). $^{13}\text{C-NMR}$: 82.0 (*t*); 51.1 (*d*); 48.0 (*t*); 47.1 (*d*); 42.1 (*s*); 33.2 (*t*); 30.5 (*s*); 29.4 (*q*); 22.4 (*q*).

trans-11: $^1\text{H-NMR}$: 4.50 (*s*, 2 H); 4.10 (*m*, 1 H); 3.18 (*d*, $J = 12.4$, 1 H); 2.32 (*dd*, $J = 8.8$, 12.0, 1 H); 1.96 (*dd*, $J = 11.5$, 12.0, 1 H); 1.95, 1.64 (*AB*, $J = 13.8$, 2 H); 1.53 (*s*, 3 H); 1.01 (*s*, 9 H). $^{13}\text{C-NMR}$: 79.9 (*t*); 57.9 (*t*); 53.0 (*d*); 46.1 (*d*); 42.1 (*s*); 35.0 (*t*); 30.5 (*s*); 29.4 (*q*); 25.4 (*q*).

trans-12: $^1\text{H-NMR}$: 4.53, 4.46 (*AB*, $J = 17.4$, 2 H); 3.67 (*m*, 1 H); 3.41 (*d*, $J = 12.2$, 1 H); 3.08 (*dd*, $J = 8.3$, 11.9, 1 H); 2.08 (*dd*, $J = 9.2$, 11.9, 1 H); 1.52 (*s*, 2 H); 1.24 (*s*, 3 H); 0.98 (*s*, 9 H). $^{13}\text{C-NMR}$: 80.0 (*t*); 53.0 (*d*); 46.2 (*d*); 46.2 (*t*); 42.1 (*s*); 34.9 (*t*); 30.5 (*s*); 29.4 (*q*); 25.2 (*q*).

3.4. Irradiation of **2** in the Presence of **4**. A soln. of **2** (114 mg) and **4** (840 mg) was irradiated for 14 h to afford, in quant. yield, a 2:3 diastereoisomer mixture of *8,8,9,9-tetramethyl-2,4-dioxabicyclo[5.2.0]nonan-6-one* (*cis-13/trans-13*). The following data all stem from the product mixture.

cis-13: 4.92, 4.84 (*AB*, $J = 4.7$, 2 H); 4.12, 3.93 (*AB*, $J = 17.7$, 2 H); 3.95 (*d*, $J = 6.3$, 1 H); 3.07 (*d*, $J = 6.3$, 1 H); 1.16, 1.08, 1.07, 0.95 (4*s*, 12 H). $^{13}\text{C-NMR}$: 211.1 (*s*); 97.1 (*t*); 79.1 (*d*); 74.5 (*t*); 57.2 (*d*); 45.0 (*s*); 44.5 (*s*); 28.2 (*q*); 26.4 (*q*); 25.2 (*q*); 24.4 (*q*).

trans-13: 5.39, 4.63 (*AB*, $J = 7.4$, 2 H); 4.33, 4.12 (*AB*, $J = 17.8$, 2 H); 3.70 (*d*, $J = 9.1$, 1 H); 3.30 (*d*, $J = 9.1$, 1 H); 1.14, 1.12, 1.03, 1.02 (4*s*, 12 H). $^{13}\text{C-NMR}$: 210.8 (*s*); 100.1 (*t*); 80.2 (*d*); 80.1 (*t*); 61.0 (*d*); 44.2 (*s*); 42.8 (*s*); 28.4 (*q*); 26.8 (*q*); 25.4 (*q*); 25.2 (*q*).

3.5. Irradiation of **2** in the Presence of **5**. A soln. of **2** (114 mg) and **4** (880 mg) was irradiated for 14 h to give, in quant. yield, a 1:4 diastereoisomer mixture of *9,9-dimethoxy-2,4-dioxabicyclo[5.2.0]nonan-6-one* (*cis-14/trans-14*). The isolation of *cis-14* is described in Sect. 4.3; the following data for *trans-14* stem from the oily product mixture: $^1\text{H-NMR}$: 5.45, 4.65 (*AB*, $J = 7.3$, 2 H); 4.50, 4.17 (*AB*, $J = 18.4$, 2 H); 3.75 (*d*, $J = 9.4$, 1 H); 3.57 (*ddd*, $J = 8.8$, 9.4, 10.9, 1 H); 3.38, 3.22 (2*s*, 6 H); 2.21 (*dd*, $J = 8.8$, 11.5, 1 H); 1.93 (*dd*, $J = 10.9$, 11.5, 1 H). $^{13}\text{C-NMR}$: 209.1 (*s*); 101.0 (*s*); 100.1 (*t*); 80.2 (*d*); 79.2 (*t*); 50.8 (*q*); 50.2 (*q*); 48.1 (*d*); 26.3 (*t*).

3.6. Irradiation of **2** in the Presence of **6**. A soln. of **2** (114 mg) and **4** (1120 mg) was irradiated for 14 h to afford, in quant. yield, a 20:5:70:5 diastereoisomer mixture of *9-(2,2-dimethylpropyl)-9-methyl-2,4-dioxabicyclo[5.2.0]nonan-6-one* (*cis-15/cis-16/trans-15/trans-16*; monitoring by GC/MS). The data of the two main products stem from this mixture.

cis-15: 4.93, 4.83 (*AB*, $J = 4.9$, 2 H); 4.25, 3.97 (*AB*, $J = 17.7$, 2 H); 4.10 (*d*, $J = 8.8$, 1 H); 3.50 (*m*, 1 H); 2.25 (*dd*, $J = 4.2$, 12.0, 1 H); 1.75 (*dd*, $J = 9.5$, 12.0, 1 H); 1.56, 1.44 (*AB*, $J = 14.1$, 2 H); 1.10 (*s*, 3 H); 0.96 (*s*, 9 H). $^{13}\text{C-NMR}$: 212.1 (*s*); 96.1 (*t*); 79.1 (*d*); 73.5 (*t*); 54.1 (*t*); 47.2 (*d*); 42.8 (*s*); 32.1 (*t*); 30.4 (*s*); 30.3 (*q*); 29.5 (*q*).

trans-**15**: 5.41, 4.57 (*AB*, $J = 7.4$, 2 H); 4.45, 4.15 (*AB*, $J = 17.8$, 2 H); 3.71 (*ddd*, $J = 8.8$, 9.1, 10.8, 1 H); 3.42 (*d*, $J = 9.1$, 1 H); 1.67–1.65 (*m*, 2 H); 1.55, 1.40 (*AB*, $J = 14.3$, 2 H); 1.35 (*s*, 3 H); 0.96 (*s*, 9 H). ^{13}C -NMR: 210.2 (*s*); 101.1 (*t*); 82.5 (*d*); 79.5 (*t*); 57.1 (*t*); 52.0 (*d*); 44.1 (*s*); 32.1 (*t*); 30.4 (*s*); 30.3 (*q*); 29.6 (*q*).

3.7. Irradiation of **3** in the Presence of **4**. A soln. of **3** (126 mg) and **4** (840 mg) was irradiated for 4 h to give, in quant. yield, a 5 : 3 diastereoisomer mixture of 3,3,7,7,8,8-hexamethyl-2-oxabicyclo[4.2.0]octan-5-one (*cis*-**17**/*trans*-**17**). The isolation of *cis*-**17** is described in Sect. 4.4; the following data for *trans*-**17** stem from the oily product mixture: ^1H -NMR: 3.82 (*d*, $J = 11.0$, 1 H); 2.65 (*d*, $J = 11.0$, 1 H); 2.30, 2.13 (*AB*, $J = 14.2$, 2 H); 1.38, 1.33, 1.18, 1.13, 1.02, 1.00 (6s, 18 H). ^{13}C -NMR: 204.2 (*s*); 80.9 (*s*); 80.1 (*d*); 59.1 (*d*); 53.8 (*t*); 44.1 (*s*); 43.8 (*s*); 27.2 (*q*); 26.4 (*q*); 24.3 (*q*); 23.2 (*q*); 22.8 (*q*); 20.2 (*q*).

3.8. Irradiation of **3** in the Presence of **5**. A soln. of **3** (126 mg) and **5** (880 mg) was irradiated for 4 h to provide, in quant. yield, a 10 : 9 diastereoisomer mixture of 8,8-dimethoxy-3,3-dimethyl-2-oxabicyclo[4.2.0]octan-5-one (*cis*-**18**/*trans*-**18**). The isolation of *cis*-**18** is described in Sect. 4.5; the following data for the *trans*-**18** stem from the oily product mixture: ^1H -NMR: 3.82 (*d*, $J = 11.0$, 1 H); 3.39, 3.28 (2s, 6 H); 2.87 (*ddd*, $J = 9.0$, 11.0, 11.1, 1 H); 2.46, 2.18 (*AB*, $J = 13.9$, 2 H); 2.15 (*dd*, $J = 9.0$, 11.2, 1 H); 1.89 (*dd*, $J = 11.0$, 11.2, 1 H); 1.46, 1.29 (2s, 6 H). ^{13}C -NMR: 204.1 (*s*); 106.0 (*s*); 83.0 (*s*); 82.9 (*d*); 54.1 (*t*); 51.9 (*q*); 51.2 (*q*); 47.5 (*d*); 31.0 (*q*); 29.2 (*t*); 27.4 (*q*).

3.9. Irradiation of **3** in the Presence of **6**. A soln. of **3** (126 mg) and **6** (1120 mg) was irradiated for 4 h to afford, in quant. yield, a 1 : 2 : 3.1 diastereoisomer mixture of 8-(2',2'-dimethylpropyl)-3,3,8-trimethyl-2-oxabicyclo[4.2.0]octan-5-one (*cis*-**19**/*cis*-**20**/*trans*-**19**/*trans*-**20**). The data of all four products stem from this product mixture.

cis-**19**: ^1H -NMR: 4.20 (*d*, $J = 6.7$, 1 H); 2.93 (*m*, 1 H); 2.57, 2.28 (*AB*, $J = 15.5$, 2 H); 2.06 (*m*, 1 H); 1.97 (*m*, 1 H); 1.54, 1.47 (*AB*, $J = 14.5$, 2 H); 1.34, 1.26, 1.10 (3s, 9 H); 0.96 (*s*, 9 H). ^{13}C -NMR: 210.1 (*s*); 79.2 (*s*); 76.5 (*d*); 53.2 (*t*); 50.5 (*t*); 42.1 (*s*); 41.5 (*d*); 37.2 (*t*); 30.9 (*q*); 30.5 (*s*); 30.4 (*q*); 27.2 (*q*); 26.5 (*q*).

cis-**20**: ^1H -NMR: 3.94 (*dd*, $J = 2.8$, 4.0, 1 H); 2.93 (*m*, 1 H); 2.50, 2.28 (*AB*, $J = 17.0$, 2 H); 2.10 (*m*, 1 H); 1.99 (*m*, 1 H); 1.59, 1.47 (*AB*, $J = 14.5$, 2 H); 1.37, 1.29, 1.16 (3s, 9 H); 0.94 (*s*, 9 H). ^{13}C -NMR: 210.1 (*s*); 79.2 (*s*); 76.9 (*d*); 50.0 (*t*); 47.1 (*t*); 41.3 (*s*); 41.1 (*d*); 38.1 (*t*); 30.9 (*q*); 30.8 (*s*); 30.4 (*q*); 26.2 (*q*); 24.5 (*q*).

trans-**19**: ^1H -NMR: 3.55 (*d*, $J = 10.7$, 1 H); 3.04 (*m*, 1 H); 2.35, 2.15 (*AB*, $J = 13.5$, 2 H); 1.67–1.58 (*m*, 2 H); 1.58, 1.46 (*AB*, $J = 14.0$, 2 H); 1.46, 1.29, 1.04 (3s, 9 H); 0.95 (*s*, 9 H). ^{13}C -NMR: 205.4 (*s*); 82.9 (*s*); 80.2 (*d*); 57.1 (*t*); 54.2 (*t*); 52.2 (*d*); 41.9 (*s*); 33.2 (*t*); 30.3 (*q*); 30.3 (*s*); 30.1 (*q*); 30.0 (*q*); 27.2 (*q*).

trans-**20**: ^1H -NMR: 3.45 (*d*, $J = 10.8$, 1 H); 3.04 (*m*, 1 H); 2.36, 2.15 (*AB*, $J = 13.5$, 2 H); 2.00 (*dd*, $J = 8.8$, 12.0, 1 H); 1.80, 1.60 (*AB*, $J = 14.0$, 2 H); 1.53 (*dd*, $J = 11.5$, 12.0, 1 H); 1.41, 1.28, 1.00 (3s, 9 H); 0.96 (*s*, 9 H). ^{13}C -NMR: 205.4 (*s*); 82.7 (*s*); 80.9 (*d*); 54.1 (*t*); 52.2 (*d*); 43.2 (*t*); 41.7 (*s*); 30.3 (*q*); 30.3 (*s*); 30.1 (*q*); 30.0 (*q*); 28.4 (*t*); 27.2 (*q*).

4. Equilibration with Basic Al_2O_3 . To a soln. of a diastereoisomer mixture of cycloadducts (0.5 mmol) in CH_2Cl_2 (5 ml) was added basic alumina (800 mg), and the suspension was stirred at r.t. for 3 h. Filtration, washing with an additional ml of CH_2Cl_2 , and evaporation of the solvent afforded one pure diastereoisomer or the correspondingly modified product mixtures (*cf.* Table).

4.1. Mixture *cis*-**8**/*trans*-**8**. Prep. TLC (pentane/ Et_2O 4 : 1) afforded 67 mg (63%) of (2*aa*,9*bb*)-1,2,2*a*,9*b*-tetrahydro-1,1,2,2-tetramethylcyclobuta[d][1]benzoxepin-3(4*H*)-one (*trans*-**8**; R_f 0.41). Light-yellow oil. ^1H -NMR: 7.35–7.05 (*m*, 4 H); 4.48, 4.37 (*AB*, $J = 17.2$, 2 H); 3.71 (*d*, $J = 13.2$, 1 H); 3.52 (*d*, $J = 13.2$, 1 H); 1.28, 1.18, 1.00, 0.96 (4s, 12 H). ^{13}C -NMR: 212.3 (*s*); 160.0 (*s*); 131.4 (*s*); 128.2 (*d*); 127.9 (*d*); 127.3 (*d*); 126.8 (*d*); 78.5 (*t*); 54.1 (*d*); 47.1 (*d*); 42.2 (*s*); 41.8 (*s*); 27.5 (*q*); 24.3 (*q*); 22.8 (*q*); 21.5 (*q*).

4.2. Mixture *cis*-**9**/*trans*-**9**. Prep. TLC (pentane/ Et_2O 1 : 1) afforded 96 mg (89%) of (2*aa*,9*ba*)-1,2,2*a*,9*b*-tetrahydro-1,1-dimethoxycyclobuta[d][1]benzoxepin-3(4*H*)-one (*cis*-**9**; R_f 0.45). Light-yellow waxy solid. M.p. 48–51°. ^1H -NMR: 7.35–7.05 (*m*, 4 H); 4.49, 4.40 (*AB*, $J = 17.1$, 2 H); 4.08 (*d*, $J = 11.0$, 1 H); 3.62 (*ddd*, $J = 6.0$, 11.0, 11.0, 1 H); 3.34, 3.10 (2s, 6 H); 3.10 (*dd*, $J = 6.0$, 12.0, 1 H); 2.37 (*dd*, $J = 11.0$, 12.0, 1 H). ^{13}C -NMR: 210.1 (*s*); 159.9 (*s*); 130.5 (*s*); 128.8 (*d*); 127.7 (*d*); 126.5 (*d*); 101.1 (*s*); 79.5 (*t*); 50.5 (*d*); 50.4 (*q*); 50.2 (*q*); 50.1 (*d*); 32.2 (*t*).

4.3. Mixture *cis*-**14**/*trans*-**14**. Prep. TLC (pentane/ Et_2O 2 : 1) afforded 88 mg (88%) of (1*a*,7*a*)-9,9-dimethoxy-2,4-dioxabicyclo[5.2.0]nonan-6-one (*cis*-**14**; R_f 0.46). White solid. M.p. 59–63°. ^1H -NMR: 5.06, 4.91 (*AB*, $J = 5.0$, 2 H); 4.39 (*d*, $J = 8.6$, 1 H); 4.31, 4.00 (*AB*, $J = 17.7$, 2 H); 3.37 (*ddd*, $J = 4.8$, 8.8,

9.8, 1 H); 3.26, 3.22 (2s, 6 H); 2.80 (*dd*, $J = 5.0, 13.1$, 1 H); 2.07 (*dd*, $J = 10.1, 13.1$, 1 H). $^{13}\text{C-NMR}$: 210.1 (s); 101.2 (s); 94.7 (t); 77.5 (d); 73.3 (t); 50.7 (q); 50.6 (q); 49.9 (d); 29.1 (t).

4.4. *Mixture cis-17/trans-17*. Prep. TLC (pentane/Et₂O 3:1) afforded 93 mg (89%) of (*1a,6a*)-3,3,7,7,8,8-hexamethyl-2-oxabicyclo[4.2.0]octan-5-one (*cis-17*; R_f 0.42). Colorless oil. $^1\text{H-NMR}$: 3.93 (d, $J = 4.8$, 1 H); 2.66 (d, $J = 4.8$, 1 H); 2.26 (s, 2 H); 1.38, 1.33, 1.18, 1.13, 1.02, 1.00 (6s, 18 H). $^{13}\text{C-NMR}$: 210.2 (s); 78.1 (s); 75.5 (d); 52.1 (t); 51.4 (d); 43.7 (s); 43.2 (s); 26.8 (q); 26.2 (q); 24.8 (q); 23.6 (q); 22.9 (q); 20.5 (q).

4.5. *Mixture cis-18/trans-18*. Prep. TLC (pentane/Et₂O 1:1) afforded 94 mg (89%) (*1a,6a*)-8,8-dimethoxy-3,3-dimethyl-2-oxabicyclo[4.2.0]octan-5-one (*cis-18*; R_f 0.45) as white solid. M.p. 56–61°. $^1\text{H-NMR}$: 4.43 (*dd*, $J = 3.0, 5.5$, 1 H); 3.28, 3.23 (2s, 6 H); 2.71 (*ddd*, $J = 5.5, 7.5, 9.0$, 1 H); 2.62, 2.25 (*AB*, $J = 17.2$, 2 H); 2.40–2.30 (*m*, 2 H); 1.40, 1.22 (2s, 6 H). $^{13}\text{C-NMR}$: 209.1 (s); 102.0 (s); 77.5 (s); 76.2 (d); 51.4 (t); 50.9 (q); 50.2 (q); 36.8 (d); 34.0 (t); 29.9 (q); 27.1 (q).

REFERENCES

- [1] J. Hehn, C. Müller, T. Bach, in ‘Handbook of Synthetic Photochemistry’, Eds. A. Albini, M. Fagnoni, Wiley-VCH, Weinheim, 2010, pp. 171–215.
- [2] P. Margaretha, in ‘Molecular and Supramolecular Photochemistry’, Eds. A. Griesbeck, J. Mattay, Marcel Dekker, New York, 2005, pp. 211–238.
- [3] A. B. Smith III, J. L. Wood, T. P. Keenan, N. Liverton, M. Visnick, *J. Org. Chem.* **1994**, *59*, 6652.
- [4] R. A. Bunce, V. L. Taylor, E. M. Holt, *J. Photochem. Photobiol., A* **1991**, *57*, 317.
- [5] H. Hart, E. Dunkelblum, *J. Org. Chem.* **1979**, *44*, 4752.
- [6] M. Quibell, A. Benn, N. Flinn, T. Monk, M. Ramjee, Y. Wang, J. Watts, *Bioorg. Med. Chem.* **2004**, *12*, 5689.
- [7] N. L. Allinger, M. Nakazaki, V. Zalkov, *J. Am. Chem. Soc.* **1959**, *81*, 4074.
- [8] K. Schmidt, P. Margaretha, *Helv. Chim. Acta* **2011**, *94*, 768.
- [9] P. Margaretha, *Liebigs Ann. Chem.* **1973**, 727.
- [10] T. B. Grindley, W. A. Szarek, *Can. J. Chem.* **1974**, *52*, 4072.
- [11] K. Schmidt, J. Kopf, P. Margaretha, *Helv. Chim. Acta* **2005**, *88*, 1922.
- [12] J.-M. Gaudin, O. Nikolaenko, J.-Y. de Saint Laumer, B. Winter, P.-A. Blanc, *Helv. Chim. Acta* **2007**, *90*, 1245.
- [13] S. W. Baldwin, M. T. Crimmins, P. M. Gross, *Tetrahedron Lett.* **1978**, *19*, 4197.
- [14] S. W. Baldwin, J. M. Wilkinson, *J. Am. Chem. Soc.* **1980**, *102*, 3634.
- [15] H. L. Gordon, S. Freeman, T. Hudlicky, *Synlett* **2005**, 2911.
- [16] S. P. Verenkin, W.-H. Peng, H.-D. Beckhaus, C. Rüchardt, *Eur. J. Org. Chem.* **1998**, 2323.
- [17] P. Kahnberg, O. Sterner, *Tetrahedron* **2001**, *57*, 7181.
- [18] E. M. Kosower, T. S. Sorensen, *J. Org. Chem.* **1963**, *28*, 687.
- [19] V. Y. Fedorenko, T. V. Bulgyna, Y. G. Shtyrilin, E. N. Klimovitskii, *Russ. J. Org. Chem.* **2004**, *40*, 1830.
- [20] E. J. Corey, G. Schmidt, *Tetrahedron Lett.* **1979**, *20*, 399.

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